

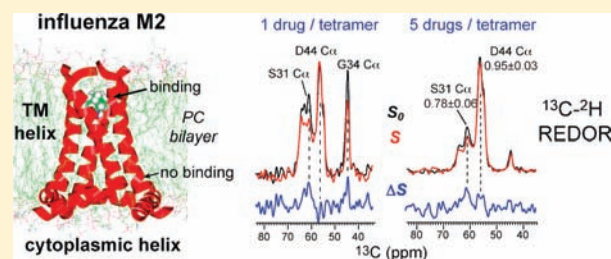
Membrane-Dependent Effects of a Cytoplasmic Helix on the Structure and Drug Binding of the Influenza Virus M2 Protein

Sarah Cady, Tuo Wang, and Mei Hong*

Department of Chemistry, Iowa State University, Ames, Iowa 50011, United States

Supporting Information

ABSTRACT: The influenza A M2 protein forms a proton channel for virus infection and also mediates virus assembly and budding. The minimum protein length that encodes both functions contains the transmembrane (TM) domain (roughly residues 22–46) for the amantadine-sensitive proton-channel activity and an amphipathic cytoplasmic helix (roughly residues 45–62) for curvature induction and virus budding. However, structural studies involving the TM domain with or without the amphipathic helix differed on the drug-binding site. Here we use solid-state NMR spectroscopy to determine the amantadine binding site in the cytoplasmic-helix-containing M2(21–61). ^{13}C – ^2H distance measurements of ^{13}C -labeled protein and ^2H -labeled amantadine showed that in 1,2-dimyristoyl-*sn*-glycero-3-phosphocholine (DMPC) bilayers, the first equivalent of drug bound S31 inside the M2(21–61) pore, similar to the behavior of M2 transmembrane peptide (M2TM) in DMPC bilayers. The nonspecific surface site of D44 observed in M2TM is disfavored in the longer peptide. Thus, the pharmacologically relevant drug-binding site in the fully functional M2(21–61) is S31 in the TM pore. Interestingly, when M2(21–61) was reconstituted into a virus-mimetic membrane containing 30% cholesterol, no chemical shift perturbation was observed for pore-lining residues, whereas M2TM in the same membrane exhibited drug-induced chemical shift changes. Reduction of the cholesterol level and the use of unsaturated phospholipids shifted the conformational equilibrium of M2TM fully to the bound state but did not rescue drug binding to M2(21–61). These results suggest that the amphipathic helix, together with cholesterol, modulates the ability of the TM helix to bind amantadine. Thus, the M2 protein interacts with the lipid membrane and small-molecule inhibitors in a complex fashion, and a careful examination of the environmental dependence of the protein conformation is required to fully understand the structure–function relation of this protein.



INTRODUCTION

The M2 protein of the influenza A virus is important for the virus lifecycle. The well-studied proton channel activity of M2 manifests itself prominently in the early stage of infection, when virus endocytosis into the acidic endosome of the host cell opens the tetrameric proton channel and acidifies the virus interior, causing the release of the ribonucleoprotein into the host cell.^{1,2} Binding of the antiviral drugs amantadine (Amt) and rimantadine (Rmt) inhibits the proton channel activity. The transmembrane (TM) domain of the protein, roughly spanning residues 22–46, is the core of the proton channel function.³ It contains the drug-binding residue S31,^{4,5} the pH-sensing and proton-selective residue H37,^{6,7} and the channel-gating residue W41.⁸ In the second function of M2, the cytoplasmic tail C-terminal to the TM domain binds the matrix protein M1 and cholesterol, and mediates virus assembly and budding.^{9,10} Electron microscopy and mutagenesis data showed that an amphipathic helix in the cytoplasmic tail, roughly spanning residues 45–62, is both sufficient and necessary for membrane scission of the newly assembled virus from the host cell.^{11,12} Simultaneous mutation of five hydrophobic residues in this amphipathic helix to Ala inhibited filamentous virion formation, virus budding, and membrane scission.^{11,12}

Although the functional role of M2 in virus assembly and budding is now clear, questions linger about the role of the amphipathic helix in M2's proton channel function. While a construct containing both the TM and amphipathic helices showed the same single-channel conductivity as the full-length protein, the TM peptide had about half the activity.³ However, the M2 transmembrane peptide (M2TM) was poorly expressed in oocytes, giving high uncertainty to the conductivity result. Indeed, the penta-Ala mutant of the full-length protein exhibited the same proton conductivity as the wild-type protein, suggesting that the amphipathic helix was not required for the proton channel activity.³ Debate about the role of the amphipathic cytoplasmic helix in the proton channel function was also fueled by two opposing high-resolution structures of drug-complexed M2. A crystal structure of Amt-bound M2TM(22–46) found Amt electron densities in the N-terminal pore, suggesting a physical occlusion mechanism for inhibition.⁵ In contrast, a solution NMR structure of Rmt-bound M2(18–60) in DHPC micelles found drug–protein nuclear Overhauser (NOE) cross peaks for residues

Received: March 6, 2011

Published: June 10, 2011

on the C-terminal lipid-facing surface of the TM helical bundle,¹³ suggesting an allosteric inhibition mechanism. A subsequent solid-state NMR study of M2TM in 1,2-dimyristoyl-*sn*-glycero-3-phosphocholine (DMPC) bilayers partly resolved this discrepancy, showing that the surface binding site was populated only by excess drugs from the membrane side, whereas the first equivalent of drug bound the N-terminal pore with much higher affinity,⁴ consistent with functional data.^{14,15} Nevertheless, it remains unclear why no pore-bound drug was detected in the solution NMR structure of M2(18–60), and the longer protein length used in that study was often cited to justify the relevance of the surface binding site.^{16,17} Recently, the orientation of M2(22–62) in lipid bilayers was determined by solid-state NMR¹⁸ and electron paramagnetic resonance constraints:¹⁹ the amphipathic helix was found to lie parallel to the membrane and pack closely against the TM helix, suggesting interactions between the two domains.

In this work, we use solid-state NMR (SSNMR) spectroscopy to answer two questions. First, does Amt bind to the pore of the fully functional cytoplasmic-helix-containing M2 construct in DMPC bilayers? Second, does the lipid bilayer composition influence the conformational equilibrium of the long peptide with respect to drug binding? We directly measured protein–drug contacts in DMPC-reconstituted M2(21–61) to test the existence of the high-affinity binding site in the pore. We also investigated M2 conformation and drug binding in two virus-mimetic membranes containing varying amounts of cholesterol, sphingomyelin (SPM), and glycerophospholipids.²⁰ Our results show that the virus-mimetic membranes shift the conformational equilibrium of the longer M2 construct to the unbound state while retaining the bound conformation of the shorter M2TM. Thus, the amphipathic helix, through a complex interplay with cholesterol, modulates the TM helix conformation to facilitate or weaken drug binding to the channel pore.

MATERIALS AND METHODS

Membrane Samples for SSNMR Experiments. Two M2 constructs M2TM (residues 22–46) and M2(21–61) were synthesized using Fmoc solid-phase chemistry (PrimmBiotech, Cambridge, MA) and purified to >95% purity. Uniformly ¹³C, ¹⁵N-labeled amino acids (Sigma-Aldrich and Cambridge Isotope Laboratories) were incorporated at residues V27, S31, G34, and D44. The first three labeled residues test the pore-binding site, whereas the labeled D44 tests the presence of the surface binding site. Most other residues implicated in surface binding by the solution NMR study¹³ showed longer distances to Rmt than did D44, and thus were not labeled. Unlabeled peptides were used for static ²H quadrupolar echo experiments to determine the number of drugs bound to the channel and the effect of membrane composition on drug binding.

The M2 peptides were reconstituted into lipid membranes by detergent dialysis. For the ¹³C, ¹⁵N-labeled peptides, the peptide–lipid molar ratios were 1:8 for M2TM and 1:15 for M2(21–61), which corresponded to similar mass ratios of ~1:2. Three types of lipid membranes were used in this study, DMPC and two mixed membranes mimicking the virus envelope lipid composition to different extents. The virus-mimetic (VM) membrane is composed of 1,2-dipalmitoyl-*sn*-glycero-3-phosphocholine (DPPC), 1,2-dipalmitoyl-*sn*-glycero-3-phosphoethanolamine (DPPE), egg SPM containing predominantly saturated palmitoyl chains, and cholesterol at a molar ratio of 21:21:28:30 (%). The modified virus-mimetic (VM+) membrane contains 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine (POPC), 1-palmitoyl-2-oleoyl-*sn*-glycero-3-1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphoethanolamine (POPE), SPM, and cholesterol at a molar ratio of 25.6:25.6:25.6:23 (%). Thus, the cholesterol mole fraction is moderately reduced in the VM+ membrane.

For the mixed membrane samples, the lipids were codissolved in chloroform and methanol and then dried under a stream of nitrogen gas to remove the bulk of the organic solvents. The film was redissolved in cyclohexane, frozen, and lyophilized to obtain a completely dry homogeneous powder. This lipid powder was suspended in a pH 7.5 phosphate buffer (10 mM Na₂HPO₄/NaH₂PO₄, 1 mM EDTA, 0.1 mM NaN₃) and freeze–thawed six times to produce a uniform vesicle suspension. The peptides were reconstituted into the lipid vesicles by dialysis using octylglucoside.²¹ The proteoliposome mixtures were centrifuged at 150000g to obtain ~40% hydrated membrane pellets, which were packed in 4 mm rotors for solid-state NMR experiments. Perdeuterated amantadine (Amt-*d*₁₅) was directly titrated into the membrane pellet. After the pellet formation and drug addition, samples for static ²H NMR experiments were lyophilized and rehydrated to ~40% with ²H-depleted water to ensure that Amt-*d*₁₅ was the only source of the ²H NMR signal. For ¹³C–²H rotational-echo-double-resonance (REDOR) experiments, Amt-*d*₁₅ was added at a ratio of 1 or 5 drugs per tetramer, corresponding to drug–lipid molar ratios of 1:60 or 1:12, respectively.

Solid-State NMR Experiments. Solid-state NMR spectra were acquired on two spectrometers (Bruker Biospin) operating at field strengths of 14.1 and 9.4 T. All static ²H experiments were conducted on the 14.1 T spectrometer at a ²H Larmor frequency of 92.12 MHz. The quadrupolar-echo experiment used a pre-echo delay of 40–50 μs, an 8 μs shorter postecho delay, and a ²H 90° pulse of 3.8 μs. The ²H time signal was left-shifted appropriately to capture the echo maximum before Fourier transformation to give spectra with a flat baseline. The spectra were measured at 303 K with 30,000–210,000 scans.

Two-dimensional (2D) ¹³C–¹³C and ¹⁵N–¹³C correlation experiments were carried out using a 4 mm ¹H/¹³C/¹⁵N magic-angle-spinning (MAS) probe. The temperatures were 243 K for DMPC samples and 273 K for virus-mimetic membrane samples. Typical MAS frequencies were 7 kHz. The 2D ¹³C–¹³C experiments used the DARR pulse sequence,²² while the 2D ¹⁵N–¹³C experiments used REDOR for polarization transfer between ¹⁵N and ¹³C.²³

¹³C-detected ²H-dephased REDOR experiments on DMPC-bound M2 peptides²⁴ were carried out using a 4 mm ¹H/¹³C/²H MAS probe on the 9.4 T NMR spectrometer operating at ¹³C and ²H Larmor frequencies of 100.71 and 61.48 MHz, respectively. The samples were spun at 4250 Hz at 243 K. At this temperature the protein was immobile but the drug remained dynamic.⁴ Most REDOR experiments involved a single selective ¹³C 180° pulse in the center of the mixing period and multiple ²H 180° pulses (12.4 μs long) spaced at half a rotor period apart. This version of REDOR removes ¹³C–¹³C scalar coupling and gives long ¹³C T₂ relaxation times to allow dipolar dephasing to be measured at long mixing times.²⁵ An alternative REDOR version containing a single ²H composite 90°90°90° pulse and multiple ¹³C hard 180° pulses⁴ was conducted for the spectrum in Figure 3b. The mixing times and number of scans used for the REDOR experiments are listed in Table S1: Supporting Information. The error bar ε for the REDOR dephasing S/S₀ was propagated from the signal-to-noise ratios (sino) of the control (S₀) and dephased (S) spectra using the equation $\epsilon = (S/S_0)(1/\text{sino}_S^2 + 1/\text{sino}_{S_0}^2)^{1/2}$.

RESULTS

Drug Binding to M2(21–61) in DMPC Bilayers. In low-melting, one-component phosphocholine membranes such as DLPC and POPC bilayers, M2TM tetramers undergo uniaxial diffusion on an intermediate time scale at ambient temperature,^{26–29} severely broadening the NMR spectra. To determine if the longer M2(21–61) retains this rotational diffusion, we measured its ¹³C spectra as a function of temperature in the DMPC bilayer, which has a main phase transition at 296 K. Figure 1 compares the ¹³C

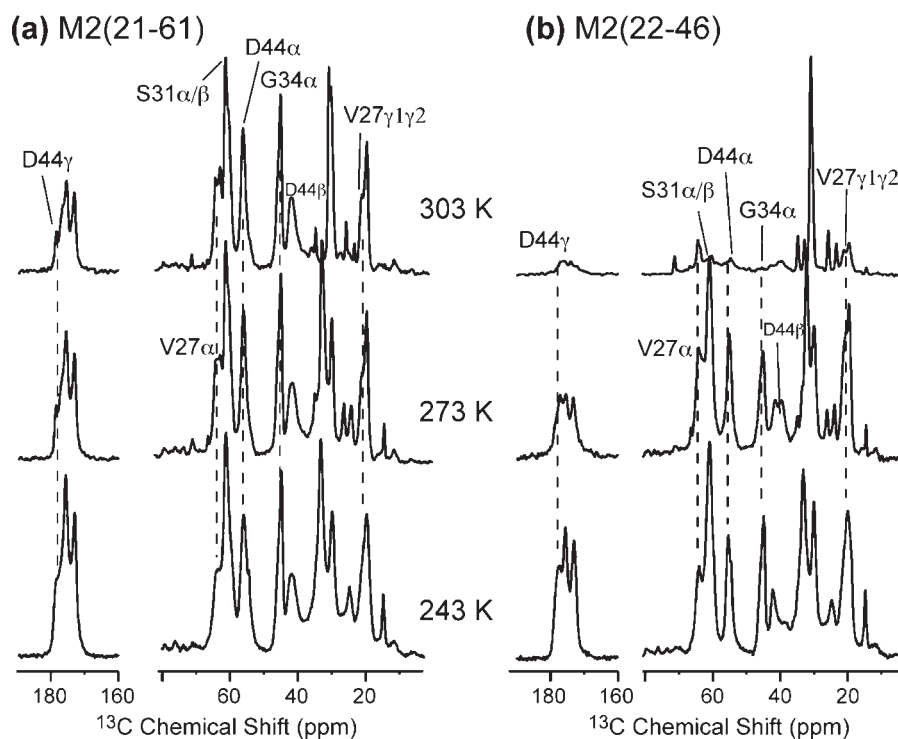


Figure 1. Variable temperature ^{13}C MAS spectra of (a) M2(21–61) and (b) M2TM(22–46) in DMPC bilayers. M2TM intensities are significantly suppressed at 303 K due to intermediate-time scale rotational diffusion of the tetramers, while M2(21–61) intensities are largely retained at 303 K, indicating immobilization of the tetramers.

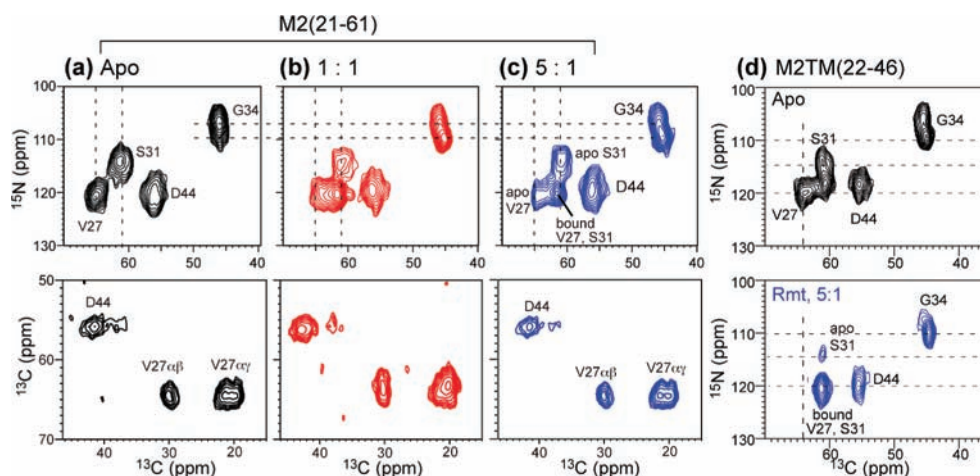


Figure 2. 2D correlation spectra of M2(21–61) and M2TM(22–46) in DMPC bilayers without and with drug. (a–c) ^{15}N – ^{13}C (top row) and ^{13}C – ^{13}C (bottom row) correlation spectra of M2(21–61) at 243 K. (a) No Amt. (b) With 1 Amt per tetramer. (c) With 5 Amt per tetramer. Note the V27 and S31 chemical shift changes and the lack of D44 chemical shift change upon drug binding. (d) 2D ^{15}N – ^{13}C correlation spectra of M2TM(22–46) without (top) and with Rmt (bottom). S31 exhibits large chemical shift changes.

cross-polarization MAS spectra of the short and long M2 constructs from 303 to 243 K. At 303 K, M2TM only exhibited 6–10% of the 243 K intensities (Figure 1b) due to uniaxial diffusion of the helical bundle, while M2(21–61) showed 60–70% of the intensities (Figure 1a). The latter is only modestly lower than the Boltzmann factor of 80% expected between 303 and 243 K, indicating that at 303 K the M2(21–61) backbone is immobilized in DMPC bilayers and only local small-amplitude motions remain. The backbone immobilization is expected because the membrane-parallel amphipathic helix significantly increases the radius of the tetrameric

assembly and should slow down rotational diffusion.³⁰ Potential interactions of the amphipathic helix with phospholipid headgroups may further restrict the tetramer mobility.

To probe how Amt affects the conformation of M2(21–61) in DMPC bilayers, we measured 2D ^{13}C – ^{13}C and ^{15}N – ^{13}C correlation spectra in the absence and presence of the drug. Amt binding to M2TM has been shown to perturb the chemical shifts of pore-lining residues, especially V27, S31, G34, and H37.^{20,28,31} Figure 2a–c shows that M2(21–61) exhibits similar chemical shift perturbations as M2(22–46): for example, the

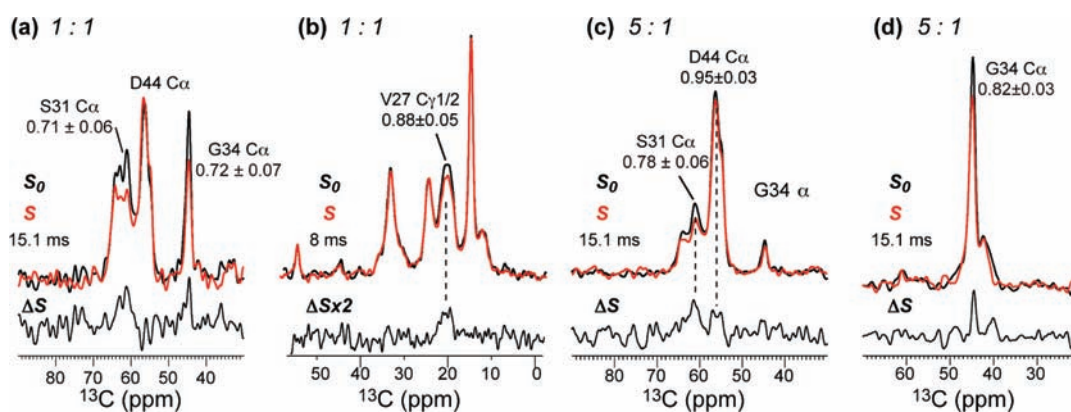


Figure 3. ^{13}C - ^2H REDOR spectra detecting the Amt binding site in DMPC-reconstituted M2(21–61). (a, b) 1 Amt per tetramer. (a) C α region showing S31 dephasing and no D44 dephasing. The spectrum was measured using the single- ^{13}C -pulse REDOR. (b) Side chain region showing V27 C γ 1 dephasing, measured using the single ^2H -pulse REDOR. (c, d) 5 Amt per tetramer. (c) C α region showing S31 dephasing and minimal D44 dephasing. (d) G34 C α region showing significant dephasing. (c, d) were measured using the single ^{13}C -pulse REDOR. S_0 and S denote control and dephased spectra measured without and with the ^2H pulses, while ΔS indicates the difference spectrum.

V27 C α peak moved from 64.1 ppm to 61.5 ppm, the S31 ^{15}N signal shifted from 114.3 ppm to 120.3 ppm, and the G34 ^{15}N peak moved from 107.6 ppm to 110.0 ppm. The similarity of the chemical shift changes suggests that Amt binds at the same location in M2(21–61) as in M2TM.^{4,21} On the other hand, the conformational equilibrium of M2(21–61) is shifted slightly toward the drug-free state compared to M2TM. At 5 drugs per tetramer, about 18% of the V27 intensity and 11% of the S31 intensity remained at the unbound positions (Figure 2c), while only ~5% of the intensities remained at the unbound positions for M2TM (Figure 2d). This difference suggests that the amphipathic helix may moderately interfere with the conformational changes of the TM helices that are necessary for drug binding. Figure 2d shows that the shorter peptide in the absence of drug already exhibits a mixed S31/V27 peak (with an ^{15}N chemical shift of 118 ppm) close to the bound position (120 ppm for ^{15}N), suggesting that the structure distribution of drug-free M2TM already contains the conformation that resembles the drug-bound state. This observation is consistent with previous studies of M2TM backbone torsion angles^{21,31} and molecular dynamics simulations of the conformational heterogeneity of M2TM in lipid bilayers.³²

Chemical shift perturbation is only an indirect indicator of the drug-binding site. To obtain definitive evidence for the drug location, we measured distance-dependent dipolar couplings between ^{13}C -labeled M2(21–61) and perdeuterated Amt. These ^{13}C - ^2H REDOR experiments were carried out under similar conditions to those of M2TM.⁴ Figure 3 shows the ^{13}C - ^2H REDOR spectra of M2(21–61) with 1 or 5 drugs per tetramer. For the 1:1 sample, we observed substantial dephasing for the S31 C α /C β peak, with a normalized intensity ratio S/S_0 of 0.71 ± 0.06 at 15 ms (Figure 3a). G34 C α and V27 C γ 1 also showed significant dephasing (Figure 3a,b), indicating that these residues constitute the boundaries of the pore binding site. No dephasing was observed for D44 at this drug concentration, indicating that the first equivalent of drug is not within atomic contact with the protein surface. These results are almost identical to those of M2TM,⁴ indicating that the amphipathic helix does not affect the drug binding behavior of M2(21–61) in DMPC bilayers under the stoichiometric drug condition.

When the drug concentration increased to 5 per tetramer, the pore-lining residues still showed REDOR dephasing

(Figure 3c,d); thus the pore binding site remains under excess drug.⁴ However, in contrast to M2TM, D44 C α in M2(21–61) exhibited little dephasing: the S/S_0 value was 0.95 ± 0.03 at 15 ms, which was insignificant. In M2TM, D44 showed a much lower S/S_0 value of 0.86 ± 0.02 already at a shorter mixing time of 10.1 ms.⁴ Thus, Amt binds minimally, if at all, to D44 in the presence of the amphipathic helix. G34 C α also retains its dephasing, but the S/S_0 value is slightly higher in the 5:1 sample than in the 1:1 sample (Figure 3d), suggesting that the height of the drug inside the pore may differ slightly between the two samples due to modulation of the TM helix packing by excess drug.²⁹ We did not attempt to quantify Amt distances to the individual residues because a significant fraction of the drug is outside the pore even at the 1:1 ratio, as shown by ^2H spectra below (Figure 4).

Amantadine Orientation and Dynamics in DMPC-Reconstituted M2(21–61). To further compare the behaviors of drug binding to M2TM versus M2(21–61), we investigated Amt orientation and dynamics using ^2H NMR. The ^2H spectra were measured at 303 K, near physiological temperature. As shown before, in protein-free DMPC bilayers, the Amt ^2H spectrum contains an 18 and 58 kHz splitting at an intensity ratio of 4:1 (Figure 4a),⁴ indicating fast uniaxial diffusion or greater than 3-site jumps of the adamantane cage around its threefold molecular axis and around the bilayer normal. The molecular axis is tilted from the bilayer normal by either 37° or 80° based on the ^2H order parameter. When a stoichiometric number of M2TM tetramers were present, a 36-kHz splitting accounted for ~76% of the total intensities (Figure 4b), indicating that most of the dynamic drugs adopt nearly upright orientation ($\sim 13^\circ$ tilt angle) due to their confinement in the channel pore.⁴ The remaining spectral intensities are located at an 18-kHz splitting (14%) and an isotropic peak (10%).⁴ When a stoichiometric number of M2(21–61) tetramers were present, the ^2H spectrum became a 47:45:7 (%) superposition of a 37-kHz splitting, the 18-kHz splitting, and an isotropic peak (Figure 4c) (simulations not shown). Therefore, only about half of all drug molecules bind to the TM pore of M2(21–61), which is a lower fraction than to M2TM. Thus, Amt is dynamic inside the immobilized M2(21–61) helical bundle. This is consistent with the fact that the 36-kHz splitting for the M2TM samples persisted from 243 to 303 K, which also supports the independence of drug motion

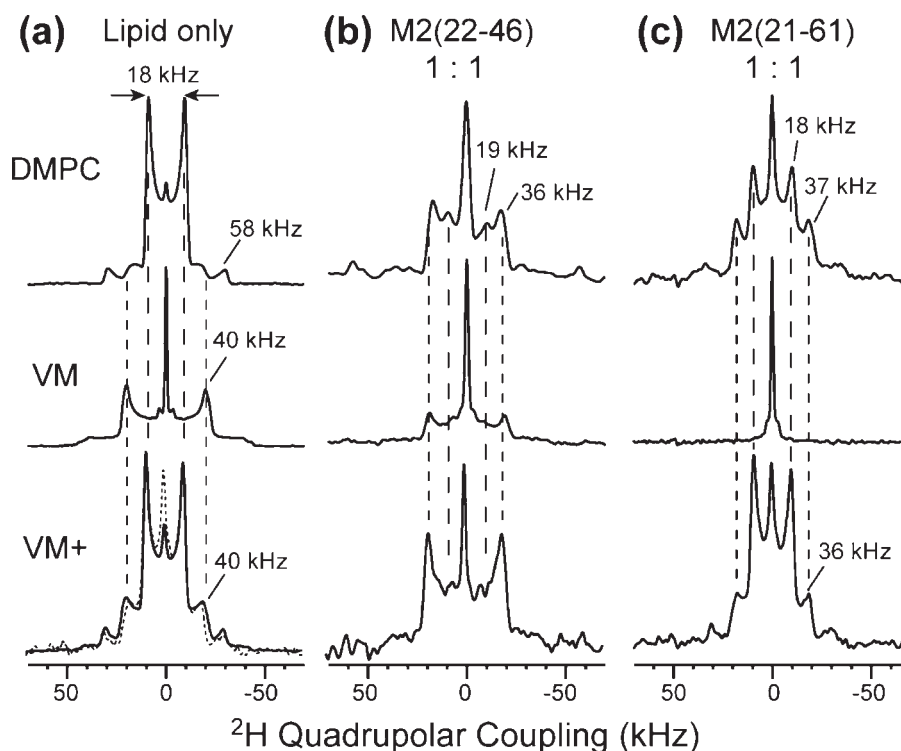


Figure 4. Static ^2H spectra of Amt- d_{15} at 303 K. The spectra were measured on DMPC bilayer samples (top row), the 30%-cholesterol VM membrane samples (middle row), and the 23%-cholesterol VM+ membrane samples (bottom row). (a) Protein-free lipid bilayers. (b) Membranes containing M2TM at a ratio of 1 drug per tetramer. (c) Membranes containing M2(21–61) at a ratio of 1 drug per tetramer.

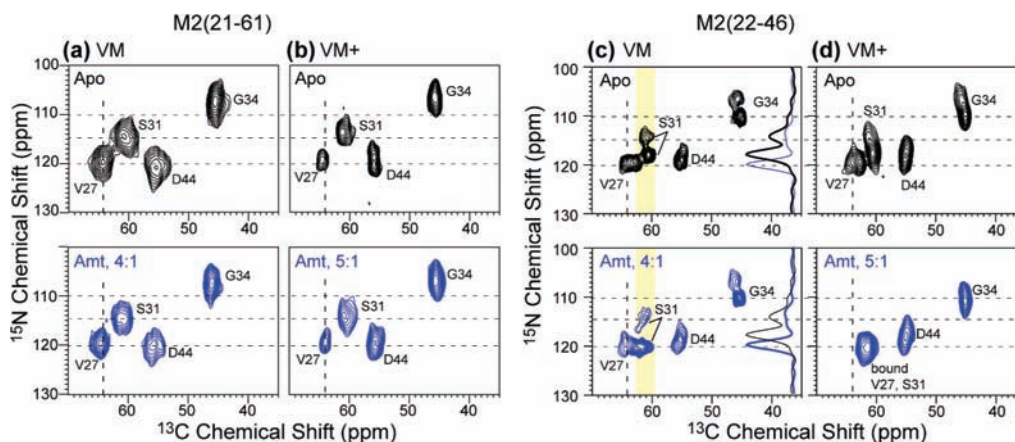


Figure 5. 2D ^{15}N – ^{13}C correlation spectra of M2(21–61) and M2TM(22–46) in cholesterol-containing membranes without (top row, black) and with drug (bottom row, blue). (a) M2(21–61) in VM membranes. (b) M2(21–61) in VM+ membranes. (c) M2TM in VM membranes. (d) M2TM in the modified VM+ membrane. The cytoplasmic-helix-containing M2(21–61) does not show chemical shift changes by Amt in either membrane, while M2TM shows chemical shift changes in both membranes and is more significantly perturbed in the VM+ membrane than in the VM membrane.

from any protein motion. The recent orientational structure of M2(22–62) showed that the TM helix in the long peptide has a similar tilt angle (32°) and packing as in M2TM,¹⁸ indicating that the pore geometry is similar between the two constructs. Thus the 18 kHz splitting in the ^2H spectrum of the M2(21–61) sample is unlikely to be due to very tilted drug molecules inside the pore, but most likely results from drugs in the lipid phase, similar to the case of the protein-free sample. Because about half of all drugs are located outside the pore and in the bilayer at this 1:1 ratio, the lack of D44 dephasing in the REDOR spectra

(Figure 3a) further supports the loss of the surface binding site in M2(21–61).

Effects of Membrane Composition on Drug Binding to M2TM and M2(21–61). The above results were obtained in DMPC bilayers, whose viscosity near the physiological temperature is intermediate between that of cholesterol-rich membranes and that of low-melting phosphocholine bilayers. Using a VM lipid mixture containing 30% cholesterol, 28% SPM, and 42% saturated DPPC and DPPE lipids,²⁰ we have reported before that M2TM exhibits similar average conformation as in

phosphocholine bilayers but different dynamics and conformational equilibrium.³¹ Because the amphipathic helix is now known to interact with cholesterol to mediate filamentous virion formation¹¹ and virus budding,¹² we investigated whether M2(21–61) interacted with Amt differently between the DMPC and VM membranes. The conformation of M2(21–61) in the VM membrane is probed through ¹³C and ¹⁵N chemical shifts and their changes upon drug binding. Figure 5a shows the 2D ¹⁵N–¹³C correlation spectra of VM-bound M2(21–61) without and with Amt. Strikingly, the chemical shifts were completely unaffected by the drug, in contrast to the shift changes of M2(21–61) in DMPC bilayers (Figure 2a,c) and the shift changes of M2TM in the VM membrane (Figure 5c). For example, in the VM membrane, drug-free M2TM showed two S31 ¹⁵N peaks at 114.5 ppm and 118 ppm with an intensity ratio of ~1:2. Amt binding shifted most of the S31 intensities to 120 ppm, similar to the situation of M2TM in DMPC bilayers.

To test whether the lack of chemical shift changes of VM-bound M2(21–61) indeed results from the lack of drug binding, we measured the ²H spectra of Amt-*d*₁₅ in the VM membrane at 303 K. In the protein-free membrane, the spectrum is dominated (~82%) by a 40 kHz splitting (Figure 4a), indicating that most drugs are upright while undergoing uniaxial rotation around the bilayer normal. This upright orientation contrasts with the 37° or 80° orientation of the drug in DMPC bilayers and can be attributed to the high viscosity of the VM membrane. The 40-kHz splitting appears to be distinct from the ~36 kHz splitting of the pore-bound drug, which corresponds to a ~13° tilt angle.⁴ The protein-free VM membrane also shows a sharp isotropic peak (18% of the total intensity) with a line width of <0.8 kHz, suggesting that a small fraction of Amt partitions into either the interlamellar aqueous solution or lipid phases with isotropic symmetry, such as micelles, small vesicles, or the cubic phase.

When a stoichiometric number of M2TM tetramers were present, the ²H spectrum was very similar to the protein-free spectrum, except that the 40-kHz splitting accounted for a smaller fraction (~75%) of the total intensity (Figure 4b). In dramatic contrast, when a stoichiometric number of M2(21–61) tetramers were present, the 40-kHz splitting disappeared altogether, leaving only the isotropic peak in the spectrum (Figure 4c), indicating that the drug no longer lies within the anisotropic environment of the lamellar bilayer. Thus, the lack of chemical shift perturbation of VM-bound M2(21–61) is correlated with the exclusion of the drug from the lamellar bilayer (Table S2). Because the same protein–lipid molar ratio (1:15) was used for the VM and DMPC samples, the absence of chemical shift changes of the VM-bound M2(21–61) cannot be attributed to the lack of protein-free membrane surface. Indeed, when the protein concentration was reduced by a factor of 2 (P/L = 1:30), we observed the same isotropic ²H spectrum (Figure S1), confirming that drug exclusion from the lamellar bilayer occurs even when there is sufficient free membrane surface.

To determine whether the loss of drug binding to M2(21–61) is due to the high viscosity of the VM membrane, we changed the VM membrane composition by replacing DPPC and DPPE with unsaturated POPC and POPE and reducing the cholesterol concentration from 30% to 23%. In the protein-free VM+ membrane, the ²H spectra of the drug resumed the 18-kHz splitting seen in DMPC bilayers (Figure 4a), indicating that the drug is able to adopt the same tilted orientation in the VM+ membrane as in the fluid DMPC bilayers. When M2TM tetramers were present at the 1:1 ratio, the ²H spectrum was a 82:11:7

(%) superposition of a 37 kHz, 14 kHz splitting and an isotropic peak (Figure 4b). Thus, the drug is predominantly upright, indicating its binding to the pore of the VM+ bound M2TM channels. This conclusion is further supported by the 2D correlation spectra, which showed clear drug-induced chemical shift changes, with almost no unbound signals of S31, V27, and G34 left (Figures Sd and S2b). In contrast, when a stoichiometric number of M2(21–61) tetramers were present in the VM+ membrane, the ²H spectrum showed only a low intensity (16% of the whole spectrum) of the 36-kHz splitting, and 2D correlation spectra exhibited very little chemical shift changes (Figures Sb and S2a). Therefore, although the fluid VM+ membrane enabled drug binding to M2TM, it did not rescue drug binding to M2(21–61) (Table S4), suggesting that cholesterol may have site-specific interactions with the amphipathic helix to affect the drug binding equilibrium to the TM pore.

DISCUSSION

The above SSNMR data not only resolve the lingering issue about how Amt interacts with the influenza M2 protein but also reveal complex interactions among the membrane composition, drug, and protein length. Direct ¹³C–²H REDOR distance data of drug-complexed M2(21–61) in DMPC bilayers show unambiguously that Amt binds S31 inside the TM pore, similar to M2TM in the same membrane. The presence of this amphipathic helix shifted the drug-binding equilibrium more toward the unbound state compared to M2TM: ²H spectra of the drug indicate that about 47% of the first equivalent of drug bound M2(21–61) while 76% of the drug bound M2TM (Figure 4b,c). The different binding equilibria are consistent with the higher remaining intensities of the unbound peaks in the spectra of the longer peptide (Figure 2c). The reduced affinity of the drug for the M2(21–61) pore compared to M2TM can be attributed to the coverage of the membrane surface by the amphipathic helix. The recent orientational study of M2(22–62) showed that the four cytoplasmic helices together cover an area of ~1000 Å².¹⁸ Given the phosphocholine headgroup area of ~65 Å², the P/L molar ratio of 1:15 used for our samples, and the two bilayer surfaces, about 50% of each bilayer surface is covered by the cytoplasmic helices. Thus, drug access to the N-terminal pore should be more restrictive for M2(21–61) than for M2TM-(22–46). Despite this reduced access, the ¹³C–²H REDOR data indicate that Amt is still able to partition into the N-terminal pore and is within atomic contact (<~6 Å) with S31, V27, and G34, while the lipid-associated drug no longer has significant contact with D44 (Figure 3c).

A recent solid-state NMR study of M2(18–60) in POPC bilayers also showed two sets of peaks that correspond to the unbound and bound conformations, of which the relative intensities depended on the Rmt concentration: with 1 Rmt per tetramer, only ~25% of the intensities lie in the bound state.¹⁶ In comparison, the equilibrium of Amt-complexed M2(21–61) in DMPC bilayers is shifted much more toward the bound state: at the 1:1 drug–tetramer ratio, 50–70% of the protein is in the bound state based on the S31 signals (Figure 2b). In addition, D44 in M2(21–61) did not exhibit noticeable chemical shift perturbation by Amt (Figure 2) but was perturbed by Rmt in the previous study. These differences cannot be attributed to chain length difference between DMPC and POPC bilayers,³³ because previous studies of the effects of the membrane thickness on M2 conformational equilibria^{20,31} indicate that the

thicker POPC bilayer should favor drug binding compared to the thinner DMPC bilayer. Instead, the Amt–Rmt binding difference may reflect a lower affinity of Rmt to the pharmacologically relevant pore binding site but a higher affinity to the surface site. This proposal is supported by the ^2H spectra of Rmt- d_{15} versus Amt- d_{15} bound to DMPC-reconstituted M2TM at the 1:1 ratio: $\sim 27\%$ of Rmt was found in the lipid-bound orientation,³⁴ while only 14% of Amt partitioned into the lipid bilayer.⁴

The REDOR spectra (Figure 3) and the drug ^2H spectra (Figure 4) of the DMPC samples indicate that Amt has little affinity to the surface D44 site in M2(21–61), in contrast to M2TM. This observation is consistent with surface plasmon resonance data of full-length M2, which showed that the drug affinity for the surface site is ~ 400 -fold weaker than for the channel pore.^{4,35} This ratio is much larger than the 40-fold affinity difference for M2TM estimated from the ^2H NMR spectra.⁴ As suggested by the orientation study of M2(22–62),¹⁸ the lower drug affinity for D44 in the longer construct is most likely due to the position and orientation of the amphipathic helix. By packing closely to the TM helix, this amphipathic helix may block drug access to D44 at the end of the TM domain. Thus, solid-state NMR data and biochemical evidence both indicate that D44 is unrelated to drug inhibition in the functionally intact M2 protein.

Given the strong affinity of Amt to the channel pore of both short and long M2 constructs in DMPC bilayers, the loss of drug binding to M2(21–61) in the VM and VM+ membranes (Figures 4 and 5) is striking and gives an example of the dramatic influence of the lipid environment on the functional structures of membrane proteins (Table S4). We attribute the absence of drug binding to a combination of the high viscosity of the VM membrane and specific interactions of cholesterol with the amphipathic helix. High viscosity of the VM membrane reduces protein conformational plasticity, as evidenced by the much larger order parameters of M2TM in VM than in phosphocholine membranes.²⁰ The importance of conformational plasticity for M2 function and drug binding has been well documented^{20,32,37–39} and is evidenced here by the fact that M2TM binds drug more fully in the fluid VM+ membrane than in the VM membrane (Figures 4 and 5).^{20,31,36–38} However, the VM+ membrane did not rescue drug binding to M2(21–61) (Figure 5a,b): the scenario that the drug may bind the pore but not cause chemical shift changes is unlikely. Thus, conformational rigidity of the protein alone is insufficient for explaining the loss of drug binding to M2(21–61). Rather, specific interactions between cholesterol and the amphipathic helix appear to be present that perturb the TM helix packing and hence the drug binding equilibrium. The N-terminal half of the cytoplasmic helix contains a cholesterol recognition amino acid consensus motif,³⁹ and M2 preparations from influenza virus-infected cells were found to contain 0.5–0.9 molecules of cholesterol per monomer. Functional studies in cholesterol-free *Escherichia coli* and cholesterol-free liposomes showed that drug-sensitive proton channel activity does not require cholesterol.³⁹ The current data suggest that not only is cholesterol not required for proton transport, but a high cholesterol level may interfere with the drug sensitivity of the channel activity. In liposomes composed of POPC, POPG and cholesterol at a 4:1:2 mol ratio, M2(18–60) exhibits Amt-sensitive proton fluxes.³ Thus, at sufficiently low cholesterol levels, the conformational equilibrium of the cytoplasmic-helix-containing M2 may be such as to allow Amt and Rmt binding to the pore. The ability of the cytoplasmic helix, through interactions with the membrane, to regulate TM helix

packing sheds light on the possible reason for the lack of pore-bound drug in the solution NMR structure of M2(18–60) in DHPC micelles.¹³ Although the detergent environment differs drastically from the virus-mimetic membranes, the cytoplasmic helix may be affected by DHPC micelles in such a way as to similarly alter the TM helix packing and abolish drug binding.

The sensitive dependence of M2 conformation and drug interaction on the lipid composition raises the question of what is the best membrane to use that reproduces the full panel of M2's functions. The current data suggest that reproducing the *average* lipid composition of the virus envelope^{40,41} may not be sufficient for M2(21–61). The spatial distribution of lipids in the virus envelope is heterogeneous, and functional studies have shown that M2 is localized to the boundary of lipid rafts,³⁹ which may have intermediate viscosities not unlike those of DMPC bilayers. Indeed, the results here indicate that DMPC bilayers are surprisingly apt at producing TM helix conformations that are competent for drug binding in both lengths of the M2 protein.³ Thus, our data validate the use of DMPC-based membranes in many solid-state NMR studies of M2TM and M2(21–61) so far.^{4,21,42} The fact that M2(21–61) does not bind Amt in the VM membrane and is unlikely to bind Amt in the VM+ membrane indicates a complex interplay between cholesterol and the amphipathic helix, which affects the structure and assembly of the TM helices. Future work directly interrogating the lipid interactions of M2(21–61) and the phase properties of cholesterol-containing membranes will be useful for elucidating the lipid-dependent conformational equilibria of the M2 protein.

■ ASSOCIATED CONTENT

S Supporting Information. Tables of REDOR experimental conditions, estimates of the fraction of bound drug, ^{31}P chemical shift principal values, and additional NMR spectra. This information is available free of charge via the Internet at <http://pubs.acs.org/>.

■ AUTHOR INFORMATION

Corresponding Author
mhong@iastate.edu

■ ACKNOWLEDGMENT

This work is supported by NIH Grant GM088204. The authors thank Dr. Jun Wang (UPenn) for providing Amt- d_{15} .

■ REFERENCES

- (1) Cady, S. D.; Luo, W. B.; Hu, F. H.; Hong, M. *Biochemistry* **2009**, *48*, 7356–7364.
- (2) Pinto, L. H.; Lamb, R. A. *J. Biol. Chem.* **2006**, *281*, 8997–9000.
- (3) Ma, C.; Polishchuk, A. L.; Ohigashi, Y.; Stouffer, A. L.; Schön, A.; Magavern, E.; Jing, X.; Lear, J. D.; Freire, E.; Lamb, R. A.; DeGrado, W. F.; Pinto, L. H. *Proc. Natl. Acad. Sci. U.S.A.* **2009**, *106*, 12283–12288.
- (4) Cady, S. D.; Schmidt-Rohr, K.; Wang, J.; Soto, C. S.; DeGrado, W. F.; Hong, M. *Nature* **2010**, *463*, 689–692.
- (5) Stouffer, A. L.; Acharya, R.; Salom, D.; Levine, A. S.; Di Costanzo, L.; Soto, C. S.; Tereshko, V.; Nanda, V.; Stayrook, S.; DeGrado, W. F. *Nature* **2008**, *451*, 596–599.
- (6) Wang, C.; Lamb, R. A.; Pinto, L. H. *Biophys. J.* **1995**, *69*, 1363–1371.
- (7) Hu, F.; Luo, W.; Hong, M. *Science* **2010**, *330*, 505–509.

- (8) Tang, Y.; Zaitseva, F.; Lamb, R. A.; Pinto, L. H. *J. Biol. Chem.* **2002**, *277*, 39880–39886.
- (9) McCown, M. F.; Pekosz, A. *J. Virol.* **2006**, *80*, 8178–8189.
- (10) Zhang, J.; Leser, G. P.; Pekosz, A.; Lamb, R. A. *Virology* **2000**, *269*, 325–334.
- (11) Rossman, J. S.; Jing, X.; Leser, G. P.; Balannik, V.; Pinto, L. H.; Lamb, R. A. *J. Virol.* **2010**, *84*, 5078–5088.
- (12) Rossman, J. S.; Jing, X.; Leser, G. P.; Lamb, R. A. *Cell* **2010**, *142*, 902–913.
- (13) Schnell, J. R.; Chou, J. J. *Nature* **2008**, *451*, 591–595.
- (14) Jing, X.; Ma, C.; Ohigashi, Y.; Oliveira, F. A.; Jardetzky, T. S.; Pinto, L. H.; Lamb, R. A. *Proc. Natl. Acad. Sci. U. S. A.* **2008**, *105*, 10967–10972.
- (15) Ohigashi, Y.; Ma, C.; Jing, X.; Balannick, V.; Pinto, L. H.; Lamb, R. A. *Proc. Natl. Acad. Sci. U. S. A.* **2009**, *106*, 18775–18779.
- (16) Andreas, L. B.; Eddy, M. T.; Pielak, R. M.; Chou, J. J.; Griffin, R. G. *J. Am. Chem. Soc.* **2010**, *132*, 10958–10960.
- (17) Pielak, R. M.; Schnell, J. R.; Chou, J. J. *Proc. Natl. Acad. Sci. U. S. A.* **2009**, *106*, 7379–7384.
- (18) Sharma, M.; Yi, M.; Dong, H.; Qin, H.; Peterson, E.; Busath, D.; Zhou, H. X.; Cross, T. A. *Science* **2010**, *330*, 509–512.
- (19) Nguyen, P. A.; Soto, C. S.; Polishchuk, A.; Caputo, G. A.; Tatko, C. D.; Ma, C.; Ohigashi, Y.; Pinto, L. H.; DeGrado, W. F.; Howard, K. P. *Biochemistry* **2008**, *47*, 9934–9936.
- (20) Luo, W.; Cady, S. D.; Hong, M. *Biochemistry* **2009**, *48*, 6361–6368.
- (21) Cady, S. D.; Mishanina, T. V.; Hong, M. *J. Mol. Biol.* **2009**, *385*, 1127–1141.
- (22) Takegoshi, K.; Nakamura, S.; Terao, T. *Chem. Phys. Lett.* **2001**, *344*, 631–637.
- (23) Hong, M.; Griffin, R. G. *J. Am. Chem. Soc.* **1998**, *120*, 7113–7114.
- (24) Gullion, T.; Schaefer, J. J. *Magn. Reson.* **1989**, *81*, 196–200.
- (25) Jaroniec, C. P.; Tounge, B. A.; Herzfeld, J.; Griffin, R. G. *J. Am. Chem. Soc.* **2001**, *123*, 3507–3519.
- (26) Song, Z.; Kovacs, F. A.; Wang, J.; Denny, J. K.; Shekar, S. C.; Quine, J. R.; Cross, T. A. *Biophys. J.* **2000**, *79*, 767–775.
- (27) Cady, S. D.; Goodman, C.; Tatko, C.; DeGrado, W. F.; Hong, M. *J. Am. Chem. Soc.* **2007**, *129*, 5719–5729.
- (28) Cady, S. D.; Hong, M. *Proc. Natl. Acad. Sci. U.S.A.* **2008**, *105*, 1483–1488.
- (29) Cady, S. D.; Hong, M. *J. Biomol. NMR* **2009**, *45*, 185–196.
- (30) Saffman, P. G.; Delbruck, M. *Proc. Natl. Acad. Sci. U.S.A.* **1975**, *72*, 3111–3113.
- (31) Hu, F.; Luo, W.; Cady, S. D.; Hong, M. *Biochim. Biophys. Acta* **2011**, *1808*, 415–423.
- (32) Yi, M.; Cross, T. A.; Zhou, H. X. *Proc. Natl. Acad. Sci. U. S. A.* **2009**, *106*, 13311–13316.
- (33) Duong-Ly, K. C.; Nanda, V.; DeGrado, W. F.; Howard, K. P. *Protein Sci.* **2005**, *14*, 856–861.
- (34) Cady, S. D.; Wang, J.; Wu, Y.; DeGrado, W. F.; Hong, M. *J. Am. Chem. Soc.* **2011**, *133*, 4274–4284.
- (35) Rosenberg, M. R.; Casarotto, M. G. *Proc. Natl. Acad. Sci. U. S. A.* **2010**, *107*, 13866–13871.
- (36) Li, C.; Qin, H.; Gao, F. P.; Cross, T. A. *Biochim. Biophys. Acta* **2007**, *1768*, 3162–3170.
- (37) Stouffer, A. L.; Ma, C.; Cristian, L.; Ohigashi, Y.; Lamb, R. A.; Lear, J. D.; Pinto, L. H.; DeGrado, W. F. *Structure* **2008**, *16*, 1067–1076.
- (38) Stouffer, A. L.; Nanda, V.; Lear, J. D.; DeGrado, W. F. *J. Mol. Biol.* **2005**, *347*, 169–179.
- (39) Schroeder, C.; Heider, H.; Möncke-Buchner, E.; Lin, T. I. *Eur. Biophys. J.* **2005**, *34*, 52–66.
- (40) Blough, H. A. *J. Gen. Virol.* **1971**, *12*, 317–320.
- (41) Klenk, H. D.; Becht, H.; Rott, R. *Virology* **1972**, *47*, 579–591.
- (42) Hu, J.; Asbury, T.; Achuthan, S.; Li, C.; Bertram, R.; Quine, J. R.; Fu, R.; Cross, T. A. *Biophys. J.* **2007**, *92*, 4335–4343.